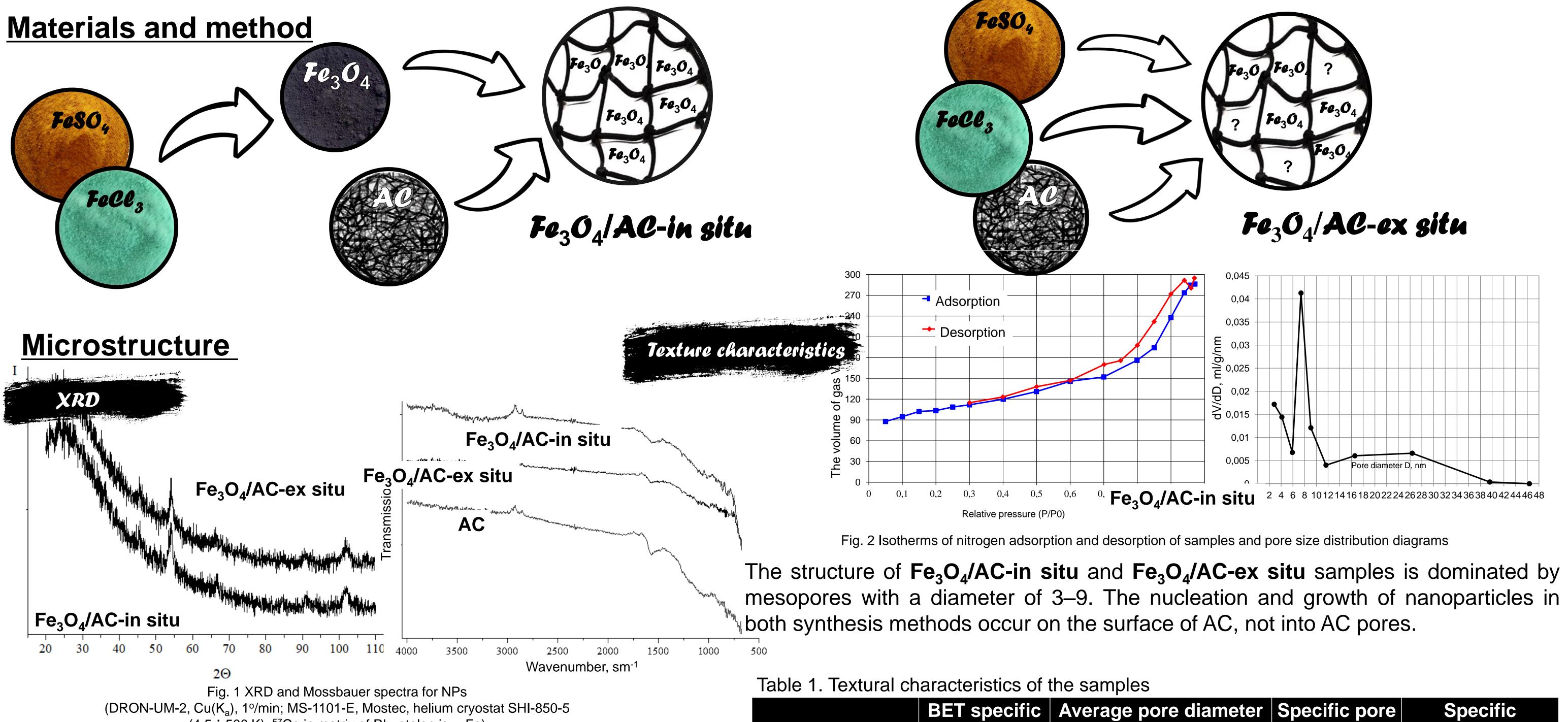
NANOCOMPOSITES BASED ON MAGNETITE AND **ACTIVATED CARBON: SYNTHESIS, SORPTION** 23th INTERNATIONAL CONFERENCE **PROPERTIES, BIOAVAILABILITY ON ENVIRONMENTAL INDICATORS**

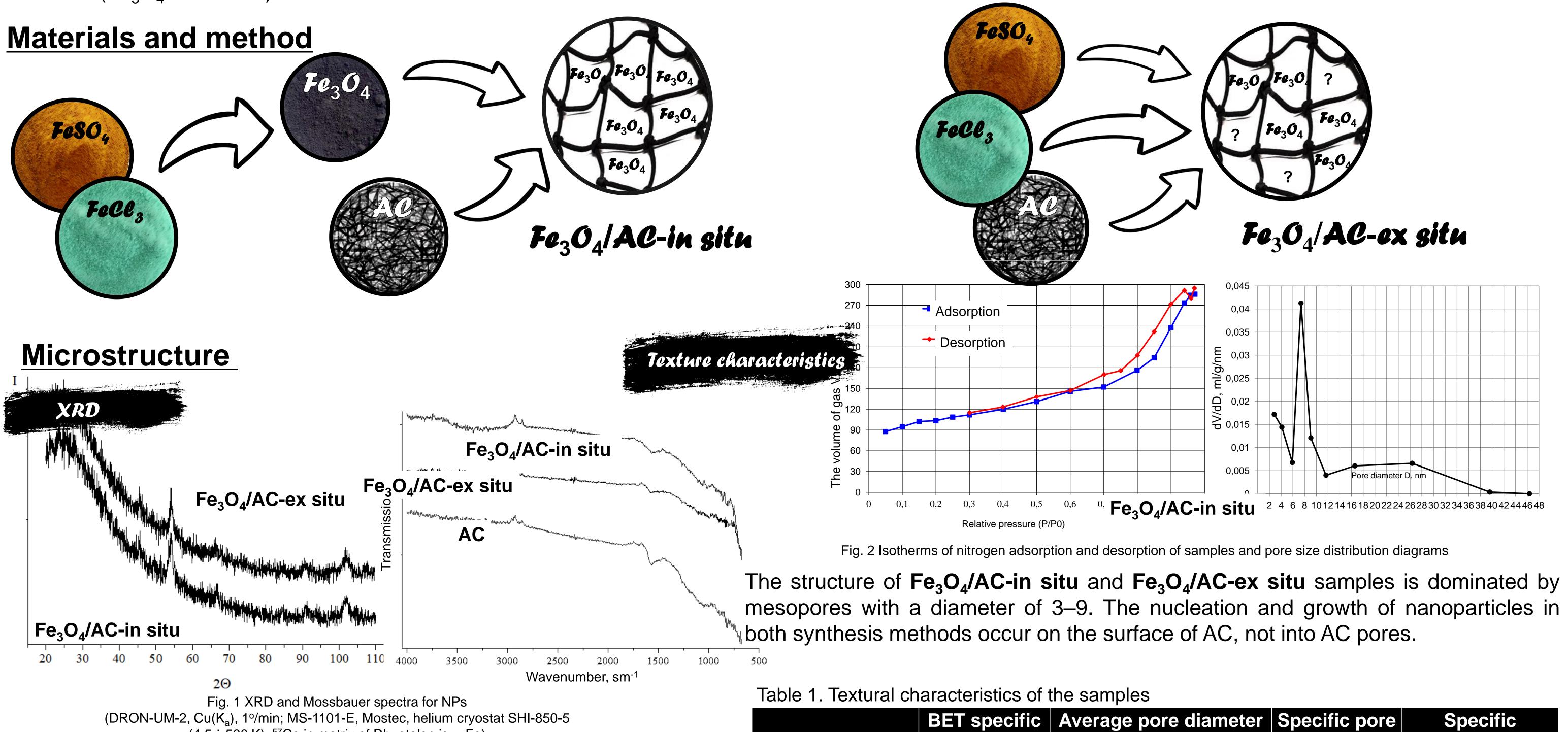


Kamila Kydralieva¹, Lyubov Bondarenko¹, Pavel Uchanov³, Vera Terekhova², Lev Rabinsky¹

¹Moscow Aviation Institute, Moscow, Russia ²Lomonosov Moscow State University, Moscow, Russia ³Severtsov Institute of Ecology and Evolution, RAS, Moscow, Russia *k_kamila@mail.ru

Hybrid organo-inorganic sorbents based on activated carbon (AC) filled with magnetically active nanoparticles provide the efficiency of purification of technogenic environment by combining the sorption properties of AC and magnetic properties of nanoparticles in order to further use the magnetic separation technique. This study was aimed to search for an effective method for the synthesis of magnetic nanocomposites based on AC and magnetite nanoparticles by chemical coprecipitation using two approaches: by preliminary synthesis of magnetite nanoparticles by chemical coprecipitation from solutions of ferrous and ferrous salts and the subsequent introduction of nanoparticles into AC matrix (Fe₃O₄-AC-ex situ) and by synthesis of magnetite nanoparticles into AC matrix (Fe₃O₄-AC-*in situ*).

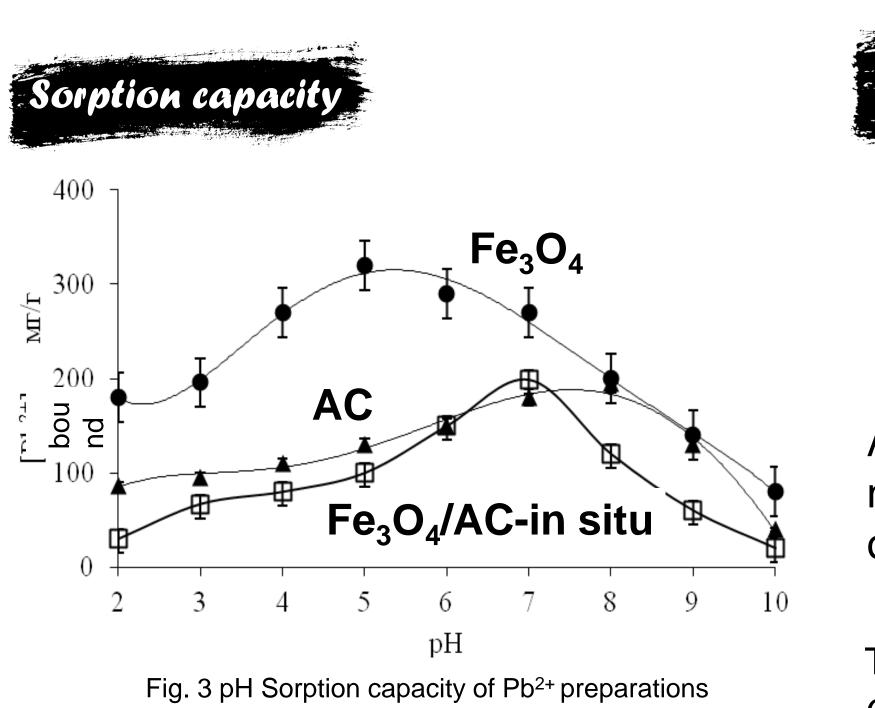




$(4.5 \div 500 \text{ K})$, ⁵⁷Co in matrix of Rh, etalon is α -Fe)

According to XRD: major phase is a magnetite Fe_3O_4 . The average particle size is 6.4 and 9.2 nm for Fe_3O_4/AC -in situ and Fe_3O_4/AC -ex **situ** samples, respectively. A weakening of the characteristic vibrational band of the carbonyl group ($Fe_3O_4/AC-ex$ situ) is observed. Simultaneously, bands corresponding to symmetric and asymmetric vibrations of carboxylate ions appear in the spectra.

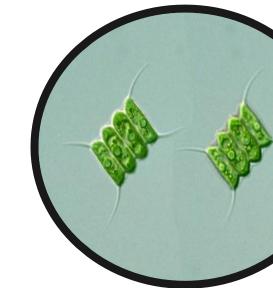
Bioactivity



The sorption activity of Fe₃O₄/AC-in *situ* reached its maximum values even at pH 5.0. The highest sorption capacity upon binding of Pb^{2+} to Fe_3O_4 nanoparticles was 320 mg/g, to AC (pH 7.5) and Fe_3O_4 /AC (pH 7) - 230 and 192 mg/g, respectively.

Sample name	surface area, m²/g	according to BJH method, nm	volume, cm³/g	micropore volume, cm ³ /g
Fe ₃ O ₄	63.8±3.8	15.8±0.9	0.2±0.01	_
AC	699.3±41.9	3.8±0.2	0.6±0.04	0.21±0.01
Fe ₃ O ₄ -AC-in situ	312.8±18.7	4.2±0.2	0.3±0.02	0.09±0.005
Fe ₃ O ₄ -AC-ex situ	259.4±15.5	4.8±0.3	0.3±0.02	0.05±0.002

The toxicity indices for the initial preparations of AC and magnetite Fe_3O_4 with respect to infusorians turned out to be similar and rather high IC50-0.078% and 0.075%, respectively. Fe_3O_4/AC -in situ showed an order of magnitude lower toxicity of IC50 - 0.17%.



According to the results of biotesting towards S. quadricauda, no high toxic effect on microalgae cells.

sharp decrease in the bioavailability of the synthesized A nanocomposites is most likely due to the microstructural characteristics of the nanocomposite.

Table 2. NOEC and EC₅₀ (%) for *Paramecium caudatum* and *Scenedesmus* quadricauda test cultures

Test function Fe₃O₄ $Fe_3O_4/AC-in situ$

The toxicity of preparations for microalgae decreases in the series

Fe_3O_4/AC in situ> $Fe_3O_4 > AC$

Probably, the inhibitory concentration of the Fe3O4/AC nanocomposite in the algological test system was affected by a more intense coloration in comparison with native magnetite and AC. The toxicity of the aqueous phase has danger to living systems. The no experiment was carried out in triplicate for each experiment in a wide range of drug concentrations: for algae - 1.0-0.001%, for infusorias - 0.01-0.0001%.

Mortality	NOEC ₁₀	EC ₅₀	NOEC ₁₀	EC_{50}	NOEC ₁₀	EC ₅₀
Paramecium caudatum	0.0003	0.075	0.0005	0.078	0.0014	0.17
Chlorophyll	NOEC ₂₀	EC ₅₀	NOEC ₂₀	EC_{50}	NOEC ₂₀	EC ₅₀
fluorescence Scenedesmus quadricauda	0.1	0.44	0.064	0.74	0.011	0.1

Conclusions

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- > Comparison of two methods for nanocomposites formulation in terms of elemental and texture characteristics of nanoparticles indicates benefits of in situ synthesis method.
- > The results of biotesting of composites towards S. quadricauda indicate the absence of a high toxic effect on microalgae cells compared to P. caudatum (IC50 in the range 0.44% - 0.10%). Based on IC50, the toxicity of preparations for microalgae decreases in the order Fe_3O_4/AC -in situ> $Fe_3O_4 > AC$.
- > Even if the sorbent is not completely removed from the medium after the magnetic separation procedure, the toxicity of the aqueous phase will not pose a danger to living systems. Acknowledgments: The study was performed with financial support of Russian Foundation for Basic Research within

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