M. V. RUDOMETKINA, YU. D. SEROPEGIN, A. V. GRIBANOV and L. S. GUSEI Department of Chemistry, Moscow Lomonosov State University, Moscow (U.S.S.R.) (Received January 13, 1988; in revised form August 2, 1988)

#### Summary

Physico-chemical analysis techniques, including studies of microstructure, X-ray phase analysis, electron microprobe, high-temperature differential thermal analysis and measurements of hardness, microhardness and specific electrical resistance, are applied to study the interaction between the components in the Ti-Nb-Ge system. The phase diagram for the Ti-Ge system has been refined. The compound  $Ti_5Ge_3$  melts congruently at 2250 K. The germanides  $Ti_6Ge_5$  and  $TiGe_2$  are formed by peritectic reactions at 1920 and 1345 K, respectively. An isothermal section of the Ti-Nb-Ge phase diagram corresponding to 1170 K is constructed and homogeneity regions for ternary niobium and titanium germanide solid solutions are reported. No ternary compounds occur in the system. A polythermal phase diagram section along the quasi-binary  $Ti_5Ge_3nNb_5Ge_3$  section is given. The mechanical and electrical properties of a number of alloys are described.

#### 1. Introduction

The purpose of this work was to study phase equilibria in the Ti-Nb-Ge system at 1170 K and some physical and chemical properties of a number of alloys involved.

The interactions between the components in the Ti–Nb and Nb–Ge binary systems have been studied in detail in refs. 1 - 5. The Ti–Ge system has mainly been studied on the titanium side [6 - 8]. The Ti–Nb–Ge phase diagram in the region 0 - 50 mass% Ge at 970 and 1270 K has been studied by Heller [7].

#### 2. Experimental details

Samples studied were alloys prepared using polycrystalline germanium of semiconductor purity (99.99%), vacuum-fusion niobium (99.9%) and titanium from titanium iodide (99.9%). Sample compositions are given in Fig. 1. The alloys were melted in an arc furnace with a non-consumable



Fig. 1. Compositions of alloys studied and isothermal section of the Ti–Nb–Ge phase diagram at 1170 K.

tungsten electrode in an argon atmosphere. Fused samples were homogenized by annealing and then quenched from 1170 K in ice-cold water.

The isothermal section of the Ti-Nb-Ge phase diagram was determined using the following physico-chemical analysis techniques: microstructural studies, X-ray phase analysis, electron microprobe, high-temperature differential thermal analysis, hardness and microhardness tests and specific resistance measurements at room temperature. The microstructures of the alloys were revealed by etching the specimens with nitric and hydrofluoric acids. X-ray phase analyses were obtained on an FR-552 focusing monochromator using Cu K $\alpha_1$  and Cu K $\alpha_{av}$  lines and germanium as the internal reference. The observed reflections were identified by comparing the patterns with the ASTM catalogue. A "Camebax" micro-analyser was employed to perform local X-ray spectral analyses of samples using titanium K $\alpha$ , germanium K $\alpha$  and niobium L $\alpha$  lines. The accuracy of the measurement was 2.5 to 3 relative per cent. High-temperature differential thermal analysis involved repeated heating of samples at a constant heating rate of  $80^{\circ}$  min<sup>-1</sup>. Tungsten-tungsten/rhenium (20 at.%) thermocouples were used to register the temperature and were calibrated against aluminium, copper, platinum, palladium and iron as reference substances. Microhardness and hardness were tested by the indentation technique using a diamond square pyramid under loads of 0.98 and 49 N, respectively. Six or more indentation marks were made for each sample. Specific resistance measurements were taken with the help of a four-probe microhead [9].

## 3. Results and discussion

The phase diagram for the Ti-Ge binary system shown in Fig. 2 was determined using the differential thermal analysis technique. Germanium



Fig. 2. Phase diagram of the Ti-Ge system.

was found to decrease the titanium polymorphic transformation temperature from 1145 to 1133 K. In the region of 0 to 5 at.% Ge the diagram was drawn using earlier data [6, 7]. A peritectoid reaction  $\beta$ -Ti + Ti<sub>5</sub>Ge<sub>3</sub>  $\rightleftharpoons$  -Ti occurs at 1133 K. The germanide Ti<sub>5</sub>Ge<sub>3</sub> and a solid solution of  $\beta$ -Ti form a eutectic at 1598 K and 15 at.% Ge. The compound Ti<sub>5</sub>Ge<sub>3</sub> melts congruently, with an open maximum at 2253 K. The system involves two peritectic equilibria: L + Ti<sub>5</sub>Ge<sub>3</sub>  $\rightleftharpoons$  Ti<sub>6</sub>Ge<sub>5</sub> (1923 K) and L + Ti<sub>6</sub>Ge<sub>5</sub>  $\rightleftharpoons$  TiGe<sub>2</sub> (1348 K). A eutectic reaction L  $\rightleftharpoons$  TiGe<sub>2</sub> + Ge occurs at 1173 K and 88 - 90 at.% Ge. The solubility of titanium in germanium is only small. Thus compounds formed in the Ti-Ge system include Ti<sub>5</sub>Ge<sub>3</sub>, Ti<sub>6</sub>Ge<sub>5</sub> and TiGe<sub>2</sub>; the compounds Ti<sub>3</sub>Ge and TiGe were not detected. The X-ray phase and electron microprobe analyses data (Table 1) are also in agreement with the occurrence of the compound Ti<sub>6</sub>Ge<sub>5</sub> mentioned in ref. 8. The 1170 K isothermal section of the Ti-Nb-Ge phase diagram is shown in Fig. 1.

Homogeneity regions of ternary solid solutions based on binary compounds occurring in the Ti-Nb-Ge system were found to extend over 10 at.% Ti for Nb<sub>3</sub>Ge, 15 - 18 at.% Ti for Nb<sub>5</sub>Ge<sub>3</sub>, 20 at.% Ti for NbGe<sub>2</sub>, 37.5 at.% Nb for Ti<sub>5</sub>Ge<sub>3</sub>, 30 at.% Nb for Ti<sub>6</sub>Ge<sub>5</sub> and 2 - 3 at.% Nb for TiGe<sub>2</sub>.

The data obtained by electron probe analysis have enabled us to refine the lines of phase equilibria for the isothermal section. Precision measurements of lattice parameters have been performed for samples corresponding to the homogeneity regions of ternary solid solutions of binary compounds. The calculation results on the ternary solid solution based on Nb<sub>3</sub>Ge are listed in Table 2, those for alloys in the TiGe<sub>2</sub>-NbGe<sub>2</sub> section in Table 3 and those for Ti<sub>5</sub>Ge<sub>3</sub>-Nb<sub>5</sub>Ge<sub>3</sub> alloys in Table 4 and Fig. 3.

### TABLE 1

Results of electron microprobe analysis of Ti-Nb-Ge alloys

Alloy c	ompositic	on (at.%)	Number of phases	Phase composition (at.%)		Phase	
Ti	Nb	Ge		Ti	Nb	Ge	
25.0	45.0	30.0	2	25.3	37.9	36.8	Ti <sub>5</sub> Ge <sub>3</sub>
				25.2	72.6	2.2	$(\beta$ -Ti + Nb)
20.0	60.0	20.0	2	13.4	67.5	19.1	Nb <sub>3</sub> Ge
				22.1	76.2	1.7	$(\beta$ -Ti + Nb)
10.0	71.0	19.0	1	12.2	67.9	19.9	Nb <sub>3</sub> Ge
10.0	30.0	60.0	2	11.0	22.0	67.0	NbGe <sub>2</sub>
				6.2	55.5	38.3	Nb <sub>5</sub> Ge <sub>3</sub>
30.0	20.0	50.0	2	33.1	22.1	44.8	Ti <sub>6</sub> Ge <sub>5</sub>
				16.9	16.6	66.5	NbGe <sub>2</sub>
20.0	30.0	50.0	2	21.3	34.9	43.8	Ti <sub>6</sub> Ge <sub>5</sub>
				16.2	16.7	67.1	NbGe <sub>2</sub>
18.3	15.0	66.7	2	3.9	28.7	67.4	TiGe <sub>2</sub>
				18.6	13.2	68.2	$NbGe_2$
42.0		58.0	2	53.7		46.3	Ti <sub>6</sub> Ge5
				34.0		66.0	TiGe <sub>2</sub>
50.0		50.0	2	33.7		66.3	TiGe <sub>2</sub>
				54.0		46.0	Ti <sub>6</sub> Ge <sub>5</sub>
55.5		44.5	1	55.0		45.0	Ti <sub>6</sub> Ge <sub>5</sub>

# TABLE 2

Physico-chemical analyses data on alloys in the region of ternary  $Nb_3Ge$  solid solutions (19 at.% Ge)

Composition (at.%)		Lattice parameters	Specific electrical	Microhardness	Hardness	
Ti	Nb	(nm)	resistance ( $\Omega$ m)	(MPa)	(MPa)	
	81.0	0.5172(1)	64 ± 5	9100 ± 100	8500 ± 100	
2.5	78.5	0.5166(1)	$82 \pm 5$	$9800 \pm 100$	8900 ± 100	
5.0	76.0	0.5265(1)	79±5	$9500 \pm 100$	$8200 \pm 100$	
7.5	73.5	0.5163(1)	69 ± 5	$8800 \pm 100$	$7000 \pm 200$	
10.0	71.0	0.5161(2)	120 ± 5	8800 ± 200	5600 ± 300	

Alloys of the  $Ti_5Ge_3$ -Nb<sub>5</sub>Ge<sub>3</sub> section were also studied by differential thermal analysis. The polythermal section shown in Fig. 4 indicates a eutectic nature of interactions between  $Ti_5Ge_3$  ( $T_m = 2253 \pm 5$  K) and Nb<sub>5</sub>Ge<sub>3</sub> ( $T_m = 2433$  K). The non-variant equilibrium is observed at 2083 K. The  $Ti_5Ge_3$ -Nb<sub>5</sub>Ge<sub>3</sub> polythermal section of the Ti-Nb-Ge phase diagram is quasi-binary.

The mechanical properties (hardness and microhardness) and specific resistance have been studied for a number of Ti-Nb-Ge alloys at room temperature. The microhardness and specific resistance values are listed in Tables 2, 3 and 4. The compound  $Nb_5Ge_3$  and its ternary solid solutions are

**TABLE 3** 

Physico-chemical analyses data on TiGe<sub>2</sub>-NbGe<sub>2</sub> alloys (66.6 at.% Ge)

Composi	tion (at.%)	Lattice parameters	t (nm)	Specific electrical	Microhardness	Hardness
Ti	Nb	a	c	resistance (52 m)	(MPa)	(MPa)
33.3		b = 0.5032(3)	0.8862(6)	$210 \pm 20$	3000 ± 100	2700 ± 100
28.3	5.0	-	1	$150 \pm 20$	$3300 \pm 100$	$2000 \pm 200$
23.3	10.0	0.4939(2)	0.6781(2)	$145 \pm 20$	$5400 \pm 200$	$3800 \pm 300$
18,3	15.0	0.4938(3)	0.6782(5)	$110 \pm 20$	$5300 \pm 200$	$4500 \pm 200$
13.3	20.0	0.4949(1)	0.6785(4)	$95 \pm 20$	$5600 \pm 100$	$4900 \pm 200$
8.3	25.0	0.4957(1)	0.6791(4)	$80 \pm 10$	$6300 \pm 200$	$5300 \pm 300$
3.3	30.0	0.4963(1)	0.6791(2)	$110 \pm 10$	$7200 \pm 100$	$6600 \pm 200$
1	33.3	0.4977(2)	0.6809(3)	$80 \pm 10$	$8400 \pm 100$	$7800 \pm 100$

Nb         Ti         a         c         resistance (M m)         (MPa)           62.5         -         1.0163(1)         0.5140(1)         190 ± 30         10400 ± 200           57.5         5.0         1.0160(2)         0.5126(2)         260 ± 50         11600 ± 100           57.5         10.0         1.0150(4)         0.5117(3)         190 ± 40         10700 ± 100           47.5         15.0         1.0144(2)         0.5117(3)         190 ± 40         10700 ± 100           47.5         15.0         1.0144(2)         0.5117(3)         190 ± 40         10700 ± 100           47.5         15.0         1.0144(2)         0.5117(3)         190 ± 40         10700 ± 100           47.5         15.0         1.0144(2)         0.5117(3)         190 ± 30         8400 ± 100           37.5         25.0         0.7642(5)         0.5265(5)         380 ± 30         8400 ± 100           37.5         35.0         0.7642(2)         0.5265(5)         380 ± 30         7800 ± 100           27.5         35.0         0.7624(2)         0.5255(3)         330 ± 30         1100           27.5         40.0         0.7628(2)         0.5255(3)         350 ± 20         111000	Composi	ition (at.%)	Lattice paramet	ters (nm)	Specific electrical	Microhardness	Hardness
$62.5$ - $1.0163(1)$ $0.5140(1)$ $190\pm 30$ $10400\pm 200$ $57.5$ $5.0$ $1.0160(2)$ $0.5117(3)$ $190\pm 40$ $10700\pm 100$ $52.5$ $110.0$ $1.0150(4)$ $0.5117(3)$ $190\pm 40$ $10700\pm 100$ $57.5$ $15.0$ $1.0144(2)$ $0.5117(3)$ $190\pm 40$ $10700\pm 100$ $47.5$ $15.0$ $1.0144(2)$ $0.5117(3)$ $190\pm 40$ $10700\pm 100$ $47.5$ $15.0$ $1.0144(2)$ $0.5111(2)$ $190\pm 30$ $8900\pm 100$ $47.5$ $20.0$ $1.0143(2)$ $0.5109(2)$ $190\pm 30$ $8400\pm 100$ $47.5$ $20.0$ $1.0143(2)$ $0.5109(2)$ $190\pm 30$ $8400\pm 100$ $47.5$ $25.0$ $0.7642(5)$ $0.5265(5)$ $380\pm 30$ $7800\pm 100$ $37.5$ $25.0$ $0.7633(3)$ $0.5256(5)$ $380\pm 30$ $7800\pm 100$ $27.5$ $37.5$ $0.7632(2)$ $0.5256(3)$ $360\pm 20$ $11100\pm 100$ $27.5$ $37.5$ $0.7668(2)$ $0.5256(3)$ $330\pm 30$ $100$ $27.5$ $410.0$ $0.7668(2)$ $0.5256(3)$ $330\pm 20$ $11100\pm 100$ $27.5$ $45.0$ $0.7668(2)$ $0.5256(3)$ $330\pm 20$ $11100\pm 100$ $27.5$ $45.0$ $0.7688(2)$ $0.5234(2)$ $300\pm 20$ $11100\pm 100$ $27.5$ $50.0$ $0.7568(2)$ $0.5234(3)$ $270\pm 40$ $11000\pm 100$ $27.5$ $0.7568(2)$ $0.5234(3)$ $0.520\pm 40$ $10100\pm 100$ $27.5$ $0.7768(2)$ $0.5234(2)$ <	Nb	Ti	ø	v	resistance (sl m)	(MPa)	(MPa)
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	62.5		1.0163(1)	0.5140(1)	$190 \pm 30$	$10400 \pm 200$	$9500 \pm 100$
$52.5$ $10.0$ $1.0150(4)$ $0.5117(3)$ $190 \pm 40$ $10700 \pm 100$ $47.5$ $15.0$ $1.0144(2)$ $0.5111(2)$ $160 \pm 30$ $8900 \pm 100$ $42.5$ $20.0$ $1.0144(2)$ $0.5111(2)$ $160 \pm 30$ $8400 \pm 100$ $42.5$ $20.0$ $1.0144(2)$ $0.5110(2)$ $8400 \pm 100$ $37.5$ $25.0$ $0.7642(5)$ $0.5265(5)$ $380 \pm 30$ $8400 \pm 100$ $37.5$ $25.0$ $  380 \pm 30$ $8300 \pm 100$ $37.5$ $35.0$ $0.7648(2)$ $0.5264(5)$ $360 \pm 30$ $7800 \pm 100$ $37.5$ $37.5$ $0.7633(3)$ $0.5256(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $37.5$ $0.7663(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7663(2)$ $0.5255(3)$ $330 \pm 30$ $1100 \pm 100$ $27.5$ $40.0$ $0.7668(2)$ $0.5225(3)$ $350 \pm 20$ $11100 \pm 100$ $27.5$ $40.0$ $0.7668(2)$ $0.52344(2)$ $350 \pm 20$ $11100 \pm 100$ $27.5$ $60.0$ $0.7588(2)$ $0.5234(3)$ $300 \pm 20$ $111000 \pm 100$ $27.5$ $60.0$ $0.7568(1)$ $0.5228(2)$ $210 \pm 40$ $1000 \pm 100$ $27.5$ $0.7568(1)$ $0.5228(2)$ $210 \pm 40$ $10100 \pm 100$ $27.5$ $0.7568(1)$ $0.5228(2)$ $210 \pm 40$ $10100 \pm 100$ $27.5$ $0.7568(2)$ $0.5228(2)$ $210 \pm 40$ $10100 \pm 100$ $27.5$ $0.7768(2)$ $0.7568(2)$ $0.5228(2)$ $0$	57.5	5.0	1.0160(2)	0.5126(2)	$260 \pm 50$	$11600 \pm 100$	$9900 \pm 100$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	52.5	10.0	1.0150(4)	0.5117(3)	$190 \pm 40$	$10700 \pm 100$	$9200 \pm 100$
$42.5$ $20.0$ $1.0143(2)$ $0.5109(2)$ $190 \pm 30$ $8400 \pm 100$ $37.5$ $25.0$ $ 0.7642(5)$ $0.5265(5)$ $380 \pm 30$ $5900 \pm 100$ $37.5$ $25.0$ $  380 \pm 30$ $5900 \pm 100$ $32.5$ $30.0$ $0.7646(2)$ $0.5264(5)$ $360 \pm 30$ $5900 \pm 100$ $32.5$ $30.0$ $0.7633(3)$ $0.5264(5)$ $360 \pm 30$ $7800 \pm 100$ $37.5$ $0.7633(3)$ $0.5258(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $37.5$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7626(2)$ $0.5255(3)$ $330 \pm 30$ $11100 \pm 100$ $27.5$ $40.0$ $0.7668(2)$ $0.52246(2)$ $350 \pm 20$ $11100 \pm 100$ $27.5$ $40.0$ $0.7668(2)$ $0.5244(2)$ $330 \pm 30$ $11100 \pm 100$ $27.5$ $60.0$ $0.7668(1)$ $0.5234(3)$ $300 \pm 20$ $11100 \pm 100$ $27.5$ $60.0$ $0.7585(2)$ $0.5234(3)$ $300 \pm 20$ $111000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5228(2)$ $210 \pm 40$ $10100 \pm 100$ $ 62.5$ $0.7563(2)$ $0.5228(2)$ $210 \pm 40$ $9500 \pm 100$	47.5	15.0	1.0144(2)	0.5111(2)	$160 \pm 30$	$8900 \pm 100$	$7300 \pm 200$
$37.5$ $25.0$ $0.7642(5)$ $0.5265(5)$ $380 \pm 30$ $5900 \pm 100$ $37.5$ $25.0$ $  380 \pm 30$ $5900 \pm 100$ $32.5$ $30.0$ $0.7646(2)$ $0.5264(5)$ $360 \pm 30$ $7800 \pm 100$ $32.5$ $30.0$ $0.7633(3)$ $0.5256(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $35.0$ $0.7633(3)$ $0.5256(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7626(2)$ $0.5255(3)$ $330 \pm 30$ $11100 \pm 100$ $27.5$ $40.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $111900 \pm 100$ $27.5$ $40.0$ $0.7608(2)$ $0.5244(2)$ $330 \pm 30$ $11100 \pm 100$ $27.5$ $50.0$ $0.77689(2)$ $0.5234(3)$ $300 \pm 20$ $111900 \pm 100$ $7.5$ $55.0$ $0.7768(1)$ $0.5234(3)$ $270 \pm 40$ $111000 \pm 100$ $2.5$ $60.0$ $0.7768(1)$ $0.5228(2)$ $210 \pm 30$ $10100 \pm 100$ $ 62.5$ $0.7768(2)$ $0.5228(2)$ $240 \pm 40$ $9500 \pm 100$	42.5	20.0	1.0143(2)	0.5109(2)	$190 \pm 30$	$8400 \pm 100$	$6400 \pm 200$
$37.5$ $25.0$ - $380 \pm 30$ $8300 \pm 100$ $32.5$ $30.0$ $0.7646(2)$ $0.5264(5)$ $360 \pm 30$ $8300 \pm 100$ $32.5$ $30.0$ $0.7633(3)$ $0.5256(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $35.0$ $0.7633(3)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $27.5$ $37.5$ $0.7625(2)$ $0.5252(3)$ $350 \pm 20$ $111100 \pm 100$ $27.5$ $40.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $111900 \pm 100$ $27.5$ $45.0$ $0.7608(2)$ $0.5244(2)$ $330 \pm 30$ $11100 \pm 100$ $17.5$ $55.0$ $0.7668(1)$ $0.5234(3)$ $300 \pm 20$ $111900 \pm 100$ $7.5$ $55.0$ $0.7768(2)$ $0.5234(3)$ $270 \pm 40$ $111000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5228(2)$ $210 \pm 30$ $11000 \pm 100$ $2.6$ $0.77663(2)$ $0.5228(2)$ $210 \pm 30$ $10100 \pm 100$ $2.6$ $0.7768(1)$ $0.5228(2)$ $240 \pm 40$ $9500 \pm 100$			0.7642(5)	0.5265(5)		$5900 \pm 100$	
32.5 $0.7646(2)$ $0.5264(5)$ $6000 \pm 100$ 32.5 $30.0$ $0.7633(3)$ $0.5258(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $35.0$ $0.7633(3)$ $0.5255(3)$ $330 \pm 30$ $7800 \pm 100$ $25.0$ $37.5$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $25.0$ $37.5$ $0.7625(2)$ $0.5255(3)$ $350 \pm 20$ $11100 \pm 100$ $22.5$ $40.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $111900 \pm 100$ $17.5$ $45.0$ $0.7608(2)$ $0.5244(2)$ $330 \pm 30$ $11900 \pm 100$ $12.5$ $50.0$ $0.77689(2)$ $0.5234(3)$ $300 \pm 20$ $111000 \pm 100$ $7.5$ $55.0$ $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $111000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5223(2)$ $210 \pm 30$ $10100 \pm 100$ $ 62.5$ $0.7563(2)$ $0.5228(2)$ $210 \pm 40$ $9500 \pm 100$	37.5	25.0	1	l	$380 \pm 30$	$8300 \pm 100$	$6100 \pm 200$
32.5 $30.0$ $0.7633(3)$ $0.5258(5)$ $360 \pm 30$ $7800 \pm 100$ $27.5$ $35.0$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $9800 \pm 100$ $25.0$ $37.5$ $0.7625(2)$ $0.5255(3)$ $330 \pm 20$ $11100 \pm 100$ $22.5$ $40.0$ $0.7608(2)$ $0.5252(3)$ $350 \pm 20$ $11100 \pm 100$ $27.5$ $45.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $111900 \pm 100$ $17.5$ $45.0$ $0.7605(1)$ $0.5244(2)$ $330 \pm 30$ $11900 \pm 100$ $12.5$ $50.0$ $0.77689(2)$ $0.5237(3)$ $300 \pm 20$ $111900 \pm 100$ $7.5$ $55.0$ $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $111000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5223(2)$ $210 \pm 30$ $10100 \pm 100$ $ 62.5$ $0.7763(2)$ $0.5228(2)$ $240 \pm 40$ $9500 \pm 100$			0.7646(2)	0.5264(5)		$6000 \pm 100$	
$27.5$ $35.0$ $0.7624(2)$ $0.5255(3)$ $330 \pm 30$ $300 \pm 100$ $25.0$ $37.5$ $0.7625(2)$ $0.5252(3)$ $360 \pm 20$ $11100 \pm 100$ $22.5$ $40.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $111500 \pm 100$ $17.5$ $45.0$ $0.7605(1)$ $0.5244(2)$ $330 \pm 30$ $11900 \pm 100$ $12.5$ $50.0$ $0.7665(1)$ $0.5244(2)$ $330 \pm 20$ $111900 \pm 100$ $12.5$ $50.0$ $0.7589(2)$ $0.5237(3)$ $300 \pm 20$ $111300 \pm 100$ $7.5$ $55.0$ $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $111000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5223(2)$ $210 \pm 30$ $10100 \pm 100$ $ 62.5$ $0.7763(2)$ $0.5228(2)$ $240 \pm 40$ $10100 \pm 100$	32.5	30.0	0.7633(3)	0.5258(5)	$360 \pm 30$	$7800 \pm 100$	$7000 \pm 100$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	27.5	35.0	0.7624(2)	0.5255(3)	$330 \pm 30$	$9800 \pm 100$	$7600 \pm 200$
22.5 $40.0$ $0.7608(2)$ $0.5246(2)$ $350 \pm 20$ $11500 \pm 100$ $17.5$ $45.0$ $0.7605(1)$ $0.5244(2)$ $330 \pm 30$ $11900 \pm 100$ $12.5$ $50.0$ $0.7589(2)$ $0.5237(3)$ $300 \pm 20$ $11300 \pm 100$ $7.5$ $55.0$ $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $11000 \pm 100$ $2.5$ $60.0$ $0.7568(1)$ $0.5223(2)$ $210 \pm 30$ $10100 \pm 100$ $ 62.5$ $0.7763(2)$ $0.5228(2)$ $240 \pm 40$ $10100 \pm 100$	25.0	37.5	0.7625(2)	0.5252(3)	$360 \pm 20$	$11100 \pm 100$	$8300 \pm 200$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	22.5	40.0	0.7608(2)	0.5246(2)	$350 \pm 20$	$11500 \pm 100$	$8600 \pm 200$
12.5 $50.0$ $0.7589(2)$ $0.5237(3)$ $300 \pm 20$ $11300 \pm 100$ 7.5 $55.0$ $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $11000 \pm 100$ 2.5 $60.0$ $0.7568(1)$ $0.52230(2)$ $210 \pm 30$ $10100 \pm 100$ - $62.5$ $0.7563(2)$ $0.5228(2)$ $240 \pm 40$ $10100 \pm 100$	17.5	45.0	0.7605(1)	0.5244(2)	$330 \pm 30$	$11900 \pm 100$	$8900 \pm 100$
7.555.0 $0.7585(2)$ $0.5234(3)$ $270 \pm 40$ $11000 \pm 100$ 2.5 $60.0$ $0.7568(1)$ $0.5230(2)$ $210 \pm 30$ $10100 \pm 100$ - $62.5$ $0.7563(2)$ $0.5228(2)$ $240 \pm 40$ $9500 \pm 100$	12.5	50.0	0.7589(2)	0.5237(3)	$300 \pm 20$	$11300 \pm 100$	$8800 \pm 100$
2.5 $60.0$ $0.7568(1)$ $0.5230(2)$ $210 \pm 30$ $10100 \pm 100$ - $62.5$ $0.7563(2)$ $0.5228(2)$ $240 \pm 40$ $9500 \pm 100$	7.5	55.0	0.7585(2)	0.5234(3)	$270 \pm 40$	$11000 \pm 100$	$8300 \pm 200$
$- 62.5  0.7563(2)  0.5228(2)  240 \pm 40  9500 \pm 100$	2.5	60.0	0.7568(1)	0.5230(2)	$210 \pm 30$	$10100 \pm 100$	$7300 \pm 100$
	1	62.5	0.7563(2)	0.5228(2)	$240 \pm 40$	$9500 \pm 100$	$6600 \pm 100$

	-INDEACES ALLONS (01.0 AL. /0 AC)
BLE 4 airs abominal analysis data on Ti Co M	SICU-CITEILICAL ALLAISES UALA ULI LISUE3-IN



Fig. 3. The dependence of crystal lattice parameters on alloy compositions for the  $Ti_5Ge_3$ -Nb<sub>5</sub>Ge<sub>3</sub> section.



Fig. 4. Polythermal  $\rm Ti_5Ge_3-Nb_5Ge_3$  section of the phase diagram.



Fig. 5. Microhardness as a function of composition for Ti<sub>5</sub>Ge<sub>3</sub>-Nb<sub>5</sub>Ge<sub>3</sub> alloys.



Fig. 6. Specific resistance as a function of composition for  $Ti_5Ge_3$ -Nb<sub>5</sub>Ge<sub>3</sub> alloys.

characterized by high microhardness values. Niobium admixtures increase the microhardness of the  $Ti_5Ge_3$ -based alloys. The concentration dependence of microhardness for  $Ti_5Ge_3$ -Nb<sub>5</sub>Ge<sub>3</sub> alloys is given in Fig. 5. Inflection points occur at phase region boundaries; in the two-phase region each phase is characterized by a constant microhardness value, while doping Nb<sub>3</sub>Ge with titanium causes an increase in microhardness. Ti-Nb binary alloys show reduced microhardness values which increase with the addition of germanium.

Shown in Fig. 6 is the specific resistance vs. composition dependence for  $Ti_5Ge_3-Nb_5Ge_3$  alloys. When the third component concentration is increased, the specific resistance of  $Ti_5Ge_3$ -based alloys increases and that of  $Nb_5Ge_3$  solid solutions decreases. Within the homogeneity regions of compounds  $NbGe_2$  and  $Nb_3Ge$ , the values of specific resistance vary insignificantly. Addition of niobium, however, enhances the resistance of the alloys in the composition region of the  $Ti_{6-x}Nb_xGe_5$  solid solutions.

246

## References

- 1 V. N. Eremenko, Titan i ego splavy, (Titanium and its alloys), Kiev, Izd. Akad. Nauk U.S.S. R., 1960.
- 2 I. I. Kornilov, Titan (Titanium), Nauka, Moscow, 1975.
- 3 J. L. Murray, Bull. Alloy Phase Diagrams, 2 (1981) 55.
- 4 J. L. Jorda, R. Flûkinger and A. Muller, J. Less-Common Met., 62 (1978) 25.
- 5 Yu. D. Seropegin, Dissertation, Moscow, 1980.
- 6 R. McQuillan, Inst. Met., 83 (1955) 485.
- 7 W. Z. Heller, Metallkunde, 64 (1973) 124.
- 8 P. Spinat, R. Fruchart and P. Herpin, Bull. Soc. Fr. Miner. Cristallogr., 93 (1970) 23.
- 9 K. K. Polnak and N. D. Chaplin, Pribory dlya nauchnyh Issledovanii, 33 (1962) 75.