

Laser Bleaching of Carbon Nanotubes Suspension in *N,N*-Dimethylformamide

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Abstract

It is shown experimentally that the multiwall carbon nanotubes (MWNTs) suspension in *N,N*-dimethylformamide (DMF) irreversibly bleaches in a wide spectrum range under a pulsed nanosecond laser radiation at the wavelength of 532 nm. According to the data of the transmission electron microscopy (TEM), the Raman spectroscopy and the reflection infrared spectrometry the bleaching is shown to be caused by MWNTs degradation, with the formation of new chemical bonds which result from the laser-induced chemical reactions between MWNTs and DMF molecules.

Keywords

Carbon nanotubes; Suspension; Photochemical reactions; Photo-induced bleaching

Introduction

Scientists have recently been studying light and nanocarbon materials interaction to acquire new knowledge to develop radically new devices for photonics and optoelectronics, e.g. graphene mode-locked laser [1] and passive laser switches based on suspensions and films of single-wall carbon nanotubes (SWNTs), permitting to obtain laser picosecond and subpicosecond pulses in the infrared region [2-8]. One can also cite the study of the laser radiation interaction with nanographite films that resulted in creation of a quick-response photodetector which can operate in a wide spectral range [9] at high temperatures [10]. A fundamentally new type of a polarization analyzer of laser radiation on the basis of this study has been designed and made [11], the principle of operation being the recording of the polarization-dependent surface photocurrents in nanographite films.

The study of laser radiation interaction with nanocarbon materials is also of interest from the point of view of production of new organic materials possessing different magnetic properties [12,13], magneto-optical light modulators developing [13], improvement of optical limiters based on nanocarbon materials [14-17]. Research in this field is also useful as an approach to locally engineer the band gap and defect densities along the SWNTs [18].

We have recently shown the semitransparent onion-like

carbon (OLC) suspension in *N,N*-dimethylformamide (DMF) to irreversibly bleach under high light fluence at the wavelength of 1064 nm, with the bleached fraction possessing pronounced diamagnetic properties [19]. We explained it by the laser-induced chemical reactions between OLC and DMF (graphene shells hydrogenation with formation of non-conducting fragments resembling condensed aromatic compounds) resulting in the transparent liquid formation. The photo-induced reactions between fullerenes and H-donor molecules in presence of photosensitizers were previously observed [20]. Besides, the reaction of SWNTs with hydrogen gas in the temperature range of 400-550°C and under the hydrogen pressure of 50 bar was reported [21]. The photo-induced chemical reactions of SWNTs with oxygen were also observed [22]. In that study it is shown that laser radiation causes pronounced bleaching and decay of photoluminescence from SWNTs on either glass or mica substrates. To put it another way and to take into account the fact that MWNTs can absorb the electromagnetic radiation in the wide wave band [23], it is logical to expect similar phenomena such as in the suspension of OLC in DMF to occur in the suspension of MWNTs in DMF under the laser radiation effect. In this paper we report on the study of the laser radiation interaction with this suspension.

Experimental Section

The MWNTs obtained by CVD technology in the reaction of the ethylene thermal catalytic decomposition on the Fe/Co catalysts were used in experiments. The MWNTs average diameter and length were 7-9 nm and 10-20 μm respectively (Figure 1a and 1b). The suspension of tubes in DMF with the MWNTs concentration of 0.015 g/l was dispersed ultrasonically (22 kHz, 300 W). This suspension was stable for 24 hours.

The optical scheme of this experiment is presented in Figure 2a. Laser beam of the YAG:Nd³⁺-laser with a frequency converter into second harmonic (the wavelength of 532 nm, the pulse duration of 17 ns) [24] passed through opened shutter (1) was focused by lens (2), with a focal distance of 100 mm, on an optical cuvette filled with the suspension under study (3). At closed shutter (1), the region of interaction between laser radiation and suspension (3) was photographed by photocamera (4). The diameter of the focused beam waist was 70 μm and the thickness of the cuvette was 1 mm.

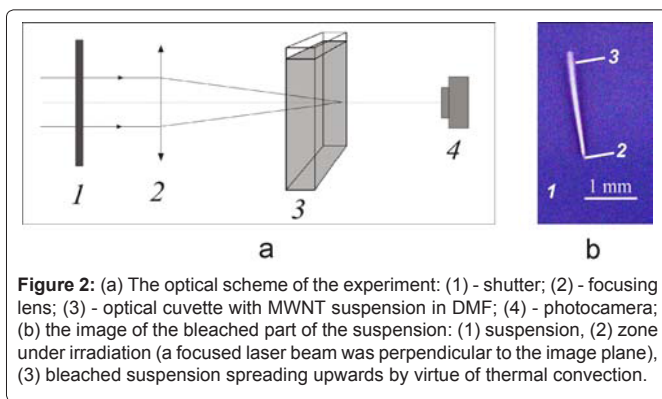
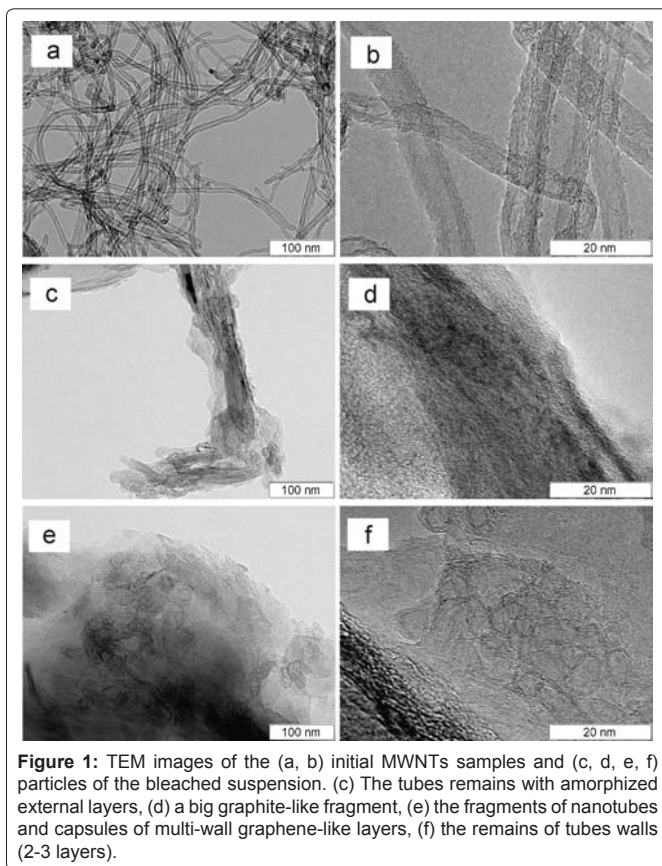
To measure the transmission spectrum of the initial and bleached suspensions, the suspensions samples were placed in a 1.01-mm quartz cuvette. The measurements were made by the PerkinElmer LAMBDA 650 double-beam UV/Vis spectrophotometer.

The transmission electron microscope JEM 2012 (JEOL, Japan) with the accelerating voltage of 200 kV was used to obtain high resolution TEM images of the MWNTs and particles of the bleached suspension. The lattice plane resolution was 0.14 nm and the point resolution was 0.194 nm.

The comparative analysis of the products contained in the initial and bleached suspensions has been carried out by the Raman spectrometer (Labram HR800, Horiba) at the wavelength of 632.8 nm and by the infrared reflectance spectrometer (FTIR 8000, Shimadzu). Thereto the films containing the solid fractions of the suspensions

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were produced on a quartz substrate by a simple exsiccation of a suspension layer deposited on the substrate.

Results and Discussion

The MWNT suspension was treated by pulsed laser radiation at different values of the input fluence W . The maximum value of the input fluence was below $2 \text{ J}\cdot\text{cm}^{-2}$. The bleaching was observed after the action of several pulses. The input fluence W of these pulses was above $0.8 \text{ J}\cdot\text{cm}^{-2}$. A typical image of the suspension bleached part is shown in Figure 2b. From this figure one can see that the bleached fraction of the suspension formed by the absorbed laser power, and therefore having a higher temperature, moves upward in the vertical direction.

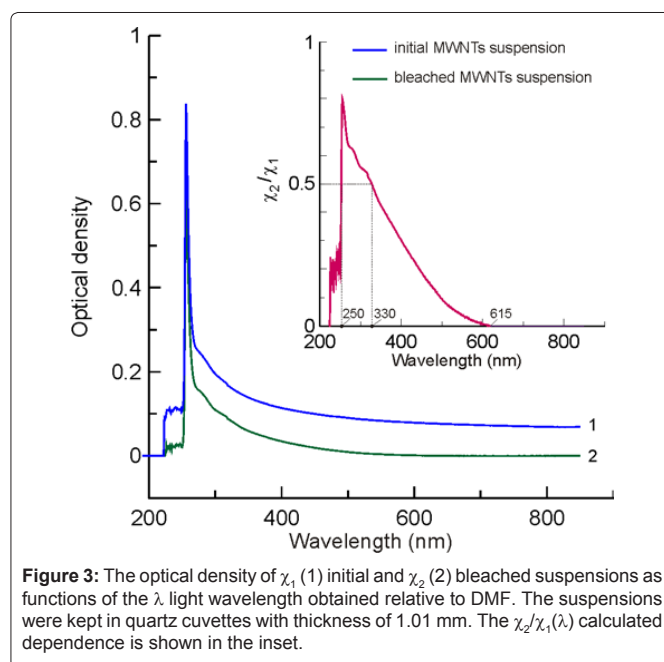
To understand and analyze the bleaching mechanisms of the

suspension, it was necessary to obtain the bleached fraction of the suspension in *quantum satis*. To this end the initial MWNTs suspension in DMF after its dispersion was injected into the optical cuvette and exposed to laser pulses for a long time, with the pulse repetition rate of 1 Hz, during $75 \times 10^3 \text{ s}$. As a result initial semitransparent suspension was transformed into a practically transparent liquid.

The bleached fraction of the suspension had a yellowish shade. The optical density of initial χ_1 (blue line) and bleached χ_2 (green line) suspensions as functions of the light wavelength (λ) obtained relative to DMF are shown in Figure 3. From these measurements one can see that the bleaching phenomenon occurs over the whole range of the wave band from 200 to 900 nm, since the inequality $\chi_1(\lambda) > \chi_2(\lambda)$ is true. However, from the calculated dependence ($\chi_2/\chi_1(\lambda)$) which is shown in the inset in Figure 3 one can conclude that the degree of bleaching depends on the wavelength. Laser bleaching reveals itself strongly over the wave bands of $330 < \lambda < 900 \text{ nm}$ and $200 < \lambda < 250 \text{ nm}$. The total bleaching was established to occur at the wavelength range of $615 < \lambda < 900 \text{ nm}$. The bleaching at the wavelength range of $250 < \lambda < 330 \text{ nm}$ is less pronounced since the χ_2/χ_1 ratio exceeds 0.5 over this wave band. It defines a weak yellowish shade of the bleached suspension obtained.

It was of interest to perform a comparative spectral analysis of materials contained in the initial and bleached suspensions using the Raman spectroscopy and the infrared reflectance spectrometry. The Raman spectrum of the initial MWNTs particles (prior to dispersion) was also obtained.

The study has shown that the Raman spectrum of the initial particles of MWNTs and samples of the MWNTs suspension practically does not differ in the frequency shifts region $1000\text{--}1800 \text{ cm}^{-1}$. The scattering spectrum of the initial MWNTs consists of two peaks with the frequency shifts of 1594.1 cm^{-1} (G band) and 1335.4 cm^{-1} (D band) (Figure 4a). The presence of the D band, with the intensity being greater than that of the G band, is associated with the presence of impurities and a lot of defects in the MWNTs studied [25].



From Figure 4b one can see that the Raman spectrum of the film obtained from the bleached fraction of the studied suspension differs considerably from that of the initial MWNTs (Figure 4a) suspension and DMF (Figure 4c). According to the reference data [26], the scattering peaks of the bleached fraction with the frequency shifts of 1074 and 1460.2 cm^{-1} are due to the CH_2 bonds, and the peak with the shift of 1138 cm^{-1} corresponds to the CH_3 bonds. The esters ($-\text{COOR}$) or aldehydes ($-\text{CHO}$) or carboxylic acids ($-\text{COOH}$) can cause a peak with the shift of 1739.9 cm^{-1} . The peak with the shift of 1613.5 cm^{-1} can be conditioned by the $-\text{NH}_2$ group. The investigations carried

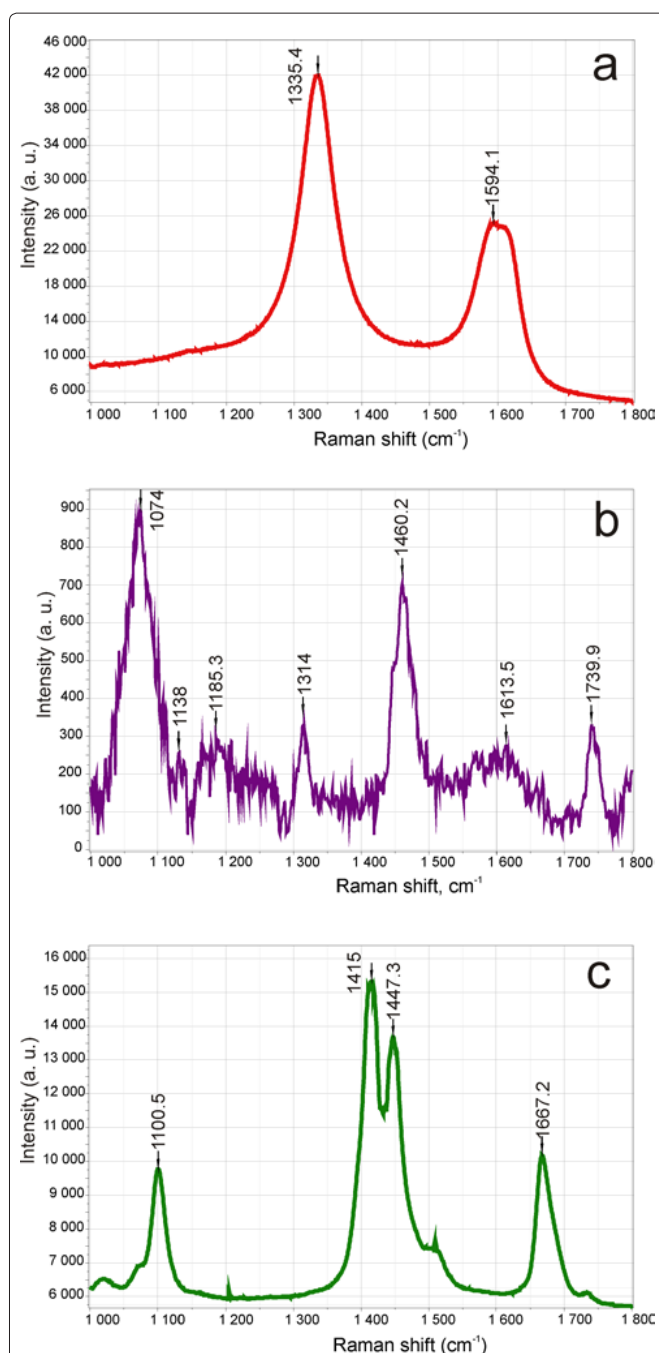


Figure 4: The Raman spectra of the films exsiccated on the quartz substrate of (a) the initial MWNTs suspension, (b) bleached MWNTs suspension, (c) The Raman spectrum of DMF.

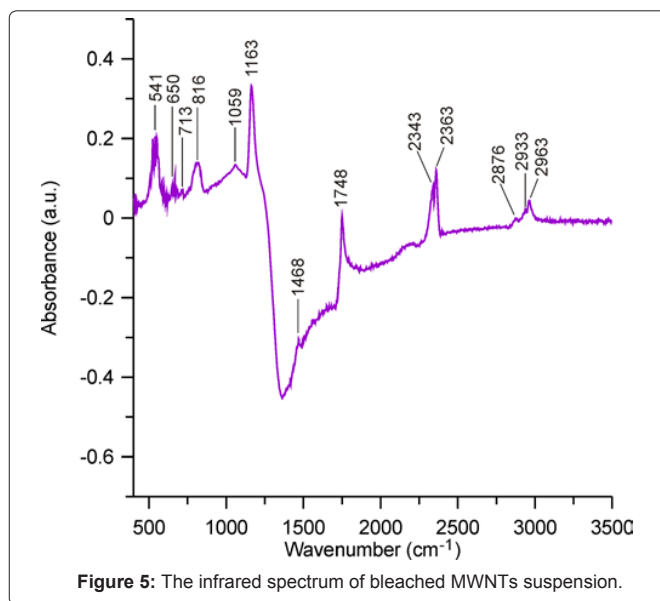


Figure 5: The infrared spectrum of bleached MWNTs suspension.

out by the infrared spectrometer also confirm the presence of the chemical bonds in the bleached fraction of the suspension (Figure 5). Thus, the peaks with the shifts of 2963 cm^{-1} , 2933 cm^{-1} , 2876 cm^{-1} , 1468 cm^{-1} correspond to the CH_3 bonds, the peaks with the shift of 1748 cm^{-1} – to the lactones, the peaks with the shifts of 1163 cm^{-1} and 1059 cm^{-1} – to the ethers (R-O-C-O-R). It is difficult to interpret the peaks with the shifts of 816 cm^{-1} , 713 cm^{-1} , 650 cm^{-1} , 541 cm^{-1} . However, one can attribute the frequency range 720-740 cm^{-1} to the methylene groups ($-\text{CH}_2-\text{CH}_2-$). The peaks with the shifts of 3400-3500 cm^{-1} and 1600 cm^{-1} were not observed in the infrared spectrum of bleached MWNTs suspension. These frequency ranges are typical for NH groups, so we can conclude that the dimethylformamide group is not present in the bleached MWNTs suspension. We should note that it was difficult to set off the effect of the background, so the peaks with the shifts of 2363 cm^{-1} and 2343 cm^{-1} correspond to CO_2 .

The TEM study of the bleached fraction of suspension has shown the bleached fraction to consist of an inhomogeneous mixture of different carbon formations (Figure 1c-1f). It can be divided into graphite-like aggregates, polyhedral thin-walled formations and amorphized carbon-bearing remains of an organic solvent. A great amount of carbon is concentrated in large (several microns) graphite-like particles consisting of bundles of graphene layers, with the thickness of 5-20 nm. In addition, the polyhedral fragmented remains of carbon nanotubes with the length of 5-20 nm are observed. The graphene walls quantity of such fragments is 2-10. The formations remotely resembling the fragments of MWNTs are also present. The organic remains of the solvent localize on the graphene layers surface and cover the polyhedral formations. The film thickness ranges from 0.5 nm to 3 nm. These data indicate that the photochemical reactions between MWNTs and DMF in the suspension of MWNTs in DMF occur under the laser exposure, which results in the formation of new chemical bonds which are not present in the initial suspension.

Conclusions

We have reported the phenomenon of the bleaching of MWNTs suspension in DMF under the powerful laser radiation effect at the wavelength of 532 nm, which reveals itself after several laser

pulses. The suspension bleaches in the wide wave band from 200 to 900 nm. The bleaching results from the photochemical reactions between MWNTs and DMF, which leads to the formation of a new stable liquid fraction. It was verified by the analysis of the initial and bleached fractions of the suspension using the transmission electron microscopy, the infrared spectrometry and the Raman spectroscopy. Although presented results demonstrate that the powerful laser irradiation destroys the graphitic structure of the MWNTs we believe that there is a proper value of laser intensity for functionalizing the MWNTs by different groups of H-donor compounds.

Acknowledgements

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